

# Low Hydroxylated Fullerenes: Stability, Thermal Behavior, and Vibrational Properties

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Extensive density functional theory calculations dedicated to analyze the stability, thermal behavior, as well as the infrared and Raman spectra of low hydroxylated fullerenols C<sub>60</sub>(OH)<sub>12</sub> are presented. It is found that the formation of local networks of hydrogenbonded OH groups plays a fundamental role in the stability of these complexes. The calculated dipole moments, polarizability values, optical gap, and Fukui functions of C<sub>60</sub>(OH)<sub>12</sub> isomers strongly depend on the structure of the hydroxyl overlayer, thus being an important parameter to tune the material properties. DFT Born–Oppenheimer molecular dynamics calculations at T = 300 K reveal that aggregated forms of OH groups on the fullerene surface show an interesting dynamical behavior, characterized by a continuous proton-exchange process between neighboring hydroxyl molecules that modifies the structure and chemical nature of the molecular coating. From nudged elastic band studies analyzing OH diffusion on the C<sub>60</sub> surface, energy barriers opposing OH migration of ~1 eV are found. However, in the presence of surrounding H<sub>2</sub>O species, a water-assisted diffusion process is obtained which can reduce the energy barriers to values as low as 0.25 eV. The comparison between experimental and calculated IR spectra of various C<sub>60</sub>(OH)<sub>12</sub> isomers shows well-defined spectral features which can be helpful to identify the structure of these fullerene complexes. Finally, simulations of the wavelength-dependent Raman spectra of hydroxylated fullerenes reveals (i) that the intensities of the Raman active modes strongly depend on the excitation laser and (ii) the importance of the wavelength-dependent calculations to reveal precise features of different regions of the spectra (J. Phys. Chem. C 2018, 122, 13117–13129).